

QUESTION 1

- 1.1. C ✓✓
- 1.2. D ✓✓
- 1.3. B ✓✓
- 1.4. A ✓✓
- 1.5. A ✓✓

[10]

QUESTION 2

- 2.1.
  - 2.1.1. A pure substance is a substance that cannot be separated into simpler components by physical methods. ✓✓
  - 2.1.2.1 Element ✓
  - 2.1.2.2 Consists of only one type of atom. ✓
  - 2.1.2.3 It is a combination of different metals (copper and zinc). ✓
- 2.2. Pots and pans are made of metal, because metal is a good conductor which allows heat to be transferred so that food can cook. ✓ The handles are insulators so that you do not burn your hands when you pick up a hot pot. ✓
- 2.3.1.
  - (a) Temperature ✓
  - (b) Time ✓
- 2.3.2. Boiling point is the temperature of a liquid at which its vapour pressure equals to the external (atmospheric) pressure ✓✓
- 2.3.3. The temperature remains constant because the energy is absorbed by the molecules. ✓  
The strength of the intermolecular forces between the molecules decreases. ✓  
The particles move faster and spaces between the particles increases. ✓  
The substance changes phase from liquid to gas ✓

[15]

# Physical Science June 2013 Chemistry Paper

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## **Physical Science June 2013 Chemistry Paper:**

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in new zealand the snares group are the closest of new zealand s subantarctic islands to mainland new zealand lying about 100 km southwest of stewart island rakiura

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islands south of invercargill wildlife havens that very few experience experience the snares auckland islands and campbell island endemic wildlife and flora are plentiful penguins seals whales megaherbs

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web 764 8 km 2 295 3 sq mi 1 coordinates 50 45 s 166 6 e 50 750 s 166 100 e 50 750 166 100 the new zealand subantarctic islands comprise the five southernmost groups of the new zealand outlying islands

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