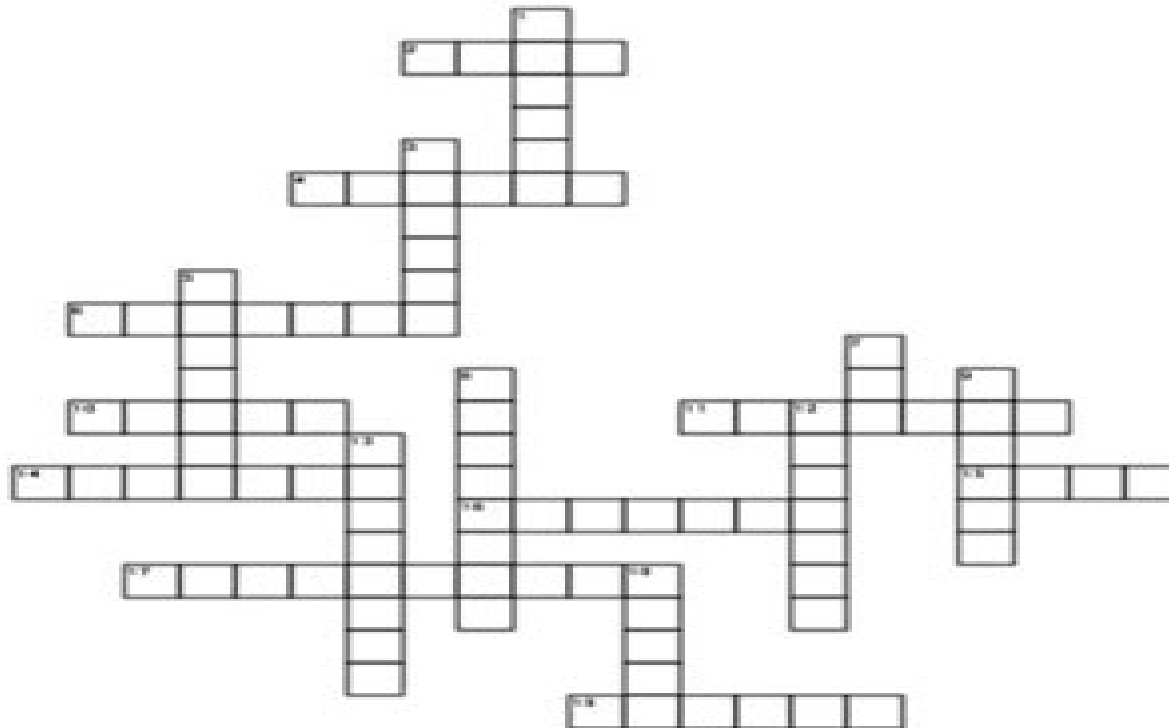


Physical Science



Across

- [2] smallest part of an element
- [4] type of paper put into a funnel to separate solids and liquids
- [6] the number of these an atom has determines what element it is
- [10] atoms vibrate
- [11] the MASS of an object divided by its VOLUME
- [14] salts dissolved in water form these when the water evaporates
- [15] on earth it is the weight of an object
- [16] charge of a neutron
- [17] educated guess
- [19] wide, flat-bottomed glass "measuring cup" with a pour spout, used in the lab

Down

- [1] the amount of space something takes up
- [3] protects hands in lab
- [5] protects eyes in labs
- [7] atoms move fast and randomly
- [8] Longer, taller container used to measure liquids
- [9] the number of protons in an atom
- [12] cluster of protons and neutrons at the center of an atom
- [13] orbits the nucleus of an atom
- [18] uses grams and measures weight

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Lei Shi



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Table of Contents Physical Science If8767 Crossword Answers

1. Understanding the eBook Physical Science If8767 Crossword Answers
 - The Rise of Digital Reading Physical Science If8767 Crossword Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science If8767 Crossword Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science If8767 Crossword Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science If8767 Crossword Answers
 - Personalized Recommendations
 - Physical Science If8767 Crossword Answers User Reviews and Ratings
 - Physical Science If8767 Crossword Answers and Bestseller Lists
5. Accessing Physical Science If8767 Crossword Answers Free and Paid eBooks
 - Physical Science If8767 Crossword Answers Public Domain eBooks
 - Physical Science If8767 Crossword Answers eBook Subscription Services
 - Physical Science If8767 Crossword Answers Budget-Friendly Options
6. Navigating Physical Science If8767 Crossword Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Physical Science If8767 Crossword Answers Compatibility with Devices
 - Physical Science If8767 Crossword Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science If8767 Crossword Answers
 - Highlighting and Note-Taking Physical Science If8767 Crossword Answers
 - Interactive Elements Physical Science If8767 Crossword Answers
 8. Staying Engaged with Physical Science If8767 Crossword Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science If8767 Crossword Answers
 9. Balancing eBooks and Physical Books Physical Science If8767 Crossword Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science If8767 Crossword Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Physical Science If8767 Crossword Answers
 - Setting Reading Goals Physical Science If8767 Crossword Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Physical Science If8767 Crossword Answers
 - Fact-Checking eBook Content of Physical Science If8767 Crossword Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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 $2 \text{ CO}_2 \text{ g} + \text{H}_2\text{O g} + \text{C}_2\text{H}_2 \text{ g} \rightarrow \text{H}_2\text{O}_2 \text{ g} + \text{C}_2\text{H}_2 \text{ g} + 2 \text{ H}_2 \text{ g} + \text{C}_2\text{H}_6 \text{ g} \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O g} + \text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g} + \frac{7}{2} \text{O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2\text{O g} \quad \Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6 \text{ g}$

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 $\text{C}_2\text{H}_4 \text{ g} + 3 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 2 \text{ H}_2\text{O l} \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6 \text{ g} + 7 \text{ O}_2 \text{ g} \rightarrow 2 \text{ CO}_2 \text{ g} + 3 \text{ H}_2\text{O l} \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2 \text{ g} + \frac{1}{2} \text{O}_2 \text{ g} \rightarrow \text{H}_2\text{O l} \quad \Delta h = 285 \text{ kJ/mole}$

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CO_2 g 2 H_2O l h 1411 kJ C_2H_6 g $3\frac{1}{2}$ O_2 g 2 CO_2 g 3 H_2O l h 1560 kJ

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