

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ relationship _____ equates _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

The _____ is a _____ reaction _____ give the _____ _____ _____ from the _____ _____ involved in the reaction _____

2. $40g = 40r + 4 \cdot 40g$

2. Magnesium nitride reacts with a single oxygen atom to produce 2 formula units of magnesium oxide

$$2 \cdot 40g = 40r + 4 \cdot 40g$$

2. $40g = 40r + 4 \cdot 40g$

$$\begin{array}{cccc} 40 & 40 & 40 & 40 \\ 40 & 40 & 40 & 40 \\ 40 & 40 & 40 & 40 \end{array}$$

2. $40g = 40r + 4 \cdot 40g$

$$2 \cdot 40g = 40r + 4 \cdot 40g$$

Note Taking Guide Episode 80stoichiometry Answers

CO Houle



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