

### Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular  $C_2H_6O$  (methylating)

Formula represents the number of \_\_\_\_\_ of \_\_\_\_\_  
in a single molecule.

Empirical  $C_2H_4O$  (methylating)

Formula represents the simplest ratio of  
positive and negative ions in a formula.

### Atomic Mass

The weight that results adding atoms together.

### Isotopes

$^{12}C$  and  $^{13}C$  isotopes

Isotopes are neutrally charged  
 $^{12}C$

### Abundance

$^{12}C$  abundance

Isotopes are neutrally charged

$^{12}C$

# Note Taking Guide Episode 601 Chemical Formula

**L Cohen**



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