natural transmutation - Uranium spontaneously decays.

$$^{238}_{92}U \rightarrow ^{234}_{90}Th + ^{4}_{2}He$$

<u>artificial transmutation</u> - bombardment of a stable isotope to force it to decay.

$$^{14}_{7}N + ^{4}_{2}He \rightarrow ^{1}_{1}p + ^{17}_{6}D$$

When the bullets are + charged, they are repelled by the nucleus they are bombarding. To overcome the repulsions, they must be outstand to very high speeds by Posticia

nuclear fission - Heavy nuclei are bombarded with neutrons and split. 1 1 ^{235}II > ^{142}D > 1 ^{91}V > 1 ^{2}I > 1 1

$$_{0}^{1}n + _{92}^{235}U \rightarrow _{56}^{142}Ba + _{36}^{91}Kr + 3(_{0}^{1}n) + E$$

Mass of particles produced is slightly <u>more</u> than the mass of the reactants.

This mass is converted into <u>energy</u>. (E = m c²)

required for a Chain sealtion

nuclear reactors: control fission <u>Chain</u> reactions to produce energy dangers:

nuclear fusion - combination of 2 nuclei into 1 with release of

$${}_{1}^{2}H + {}_{1}^{3}H \rightarrow {}_{2}^{4}He + {}_{0}^{1}n + E$$

Mass of particles produced is much MOLE than the mass of the ractions.

This Mass is converted into energy. (E = Mac)

On back, list advantages of and problems with using fusion as an energy source.

CHEMISTRY: A Study of Matter

The issues with using fusion as energy generally just come from concern for the public and their safety due to storage of nuclear material for a long time causing concerns well as radioactivity being a big issue.

Note Taking Guide Chemistry Episode 110answers

Rachel Sandford

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