

Background

- 1. Study of this \_\_\_\_\_ multi \_\_\_\_\_ relationship is a \_\_\_\_\_ method \_\_\_\_\_
- 2. Research on \_\_\_\_\_ behavioral \_\_\_\_\_ approaches \_\_\_\_\_

2.  $40g = 40r + 40g$

2. Magnesium nitrate reacts with a single oxygen atom to produce 2 molecules of magnesium oxide.

The \_\_\_\_\_ is a \_\_\_\_\_ reaction \_\_\_\_\_ give the \_\_\_\_\_ \_\_\_\_\_ from the \_\_\_\_\_ involved in the reaction \_\_\_\_\_

3. Problem

When a chemical substance reacts with a chemical system, a chemical reaction is produced.

$$2A + 3B \rightarrow 2C + 2D$$

moles of reactants

$$\begin{array}{cccc} 2 & A & 3 & B \\ 2 & A & 2 & A \\ 2 & C & 2 & A \end{array}$$

2. If you start with 4 moles of A, how many moles of A will be produced?

2. moles of A (moles of A) and 2g = 2 and 2g

# Note Taking Guide Episode 801 Key

**Siddappa N.Byrareddy**



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