reaction rate:

- the change in _____ of the _____ per unit time
- speed at which _____ disappear and _____ appear

The study of reaction rates is called _____ _____.

- ✓ In order for a reaction to occur, reacting _____ must _____ effectively.
- ✓ In an _____ collision, the _____ particles collide with the correct _____ and the proper amount of _____.

Factors Affecting Reaction Rates:

a. _____ of reactants-

- The _____ of the reactants, types of _____ involved, and state of _____ all affect reaction rates.

b. _____ of reactants-

- Increasing the concentration of reactants → more _____ collisions → increased _____ of reaction
- Increasing the _____ of the reactants (_____ only) → increases the _____ of reactants → increases _____ _____.

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Puja Mehta

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