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 using the equation from problem 2 above answer the following

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 using the following equation $2 \text{ NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2 \text{ H}_2\text{O} + \text{Na}_2\text{SO}_4$

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web google classroom balance the following chemical equation $\text{Mg} + \text{OH}^- \rightarrow \text{Mg}^{2+} + \text{OH}^-$ $\text{HCl} \rightarrow \text{HCl}$ $\text{MgCl}_2 \rightarrow \text{MgCl}_2$ $\text{H}_2\text{O} \rightarrow \text{H}_2\text{O}$ note all reactants and products require a coefficient of at least one stuck review related articles videos or use a hint report a problem 7 4 1 x x y y theta θ pi π 8 5 2 0 9 6

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