

| Element    | symbol | Atomic number | Electronic arrangement in different orbit |   |   |   |   |   |
|------------|--------|---------------|-------------------------------------------|---|---|---|---|---|
|            |        |               | K                                         | L | M | N | O | P |
| Hydrogen   | H      | 1             | 1                                         |   |   |   |   |   |
| Helium     | He     | 2             | 2                                         |   |   |   |   |   |
| Lithium    | Li     | 3             | 2                                         | 1 |   |   |   |   |
| Beryllium  | Be     | 4             | 2                                         | 2 |   |   |   |   |
| Boron      | B      | 5             | 2                                         | 3 |   |   |   |   |
| Carbon     | C      | 6             | 2                                         | 4 |   |   |   |   |
| Nitrogen   | N      | 7             | 2                                         | 5 |   |   |   |   |
| Oxygen     | O      | 8             | 2                                         | 6 |   |   |   |   |
| Fluorine   | F      | 9             | 2                                         | 7 |   |   |   |   |
| Neon       | Ne     | 10            | 2                                         | 8 |   |   |   |   |
| Sodium     | Na     | 11            | 2                                         | 8 | 1 |   |   |   |
| Magnesium  | Mg     | 12            | 2                                         | 8 | 2 |   |   |   |
| Aluminium  | Al     | 13            | 2                                         | 8 | 3 |   |   |   |
| Silicon    | Si     | 14            | 2                                         | 8 | 4 |   |   |   |
| Phosphorus | P      | 15            | 2                                         | 8 | 5 |   |   |   |
| Sulphur    | S      | 16            | 2                                         | 8 | 6 |   |   |   |
| Chlorine   | Cl     | 17            | 2                                         | 8 | 7 |   |   |   |
| Argon      | Ar     | 18            | 2                                         | 8 | 8 |   |   |   |
| Potassium  | K      | 19            | 2                                         | 8 | 8 | 1 |   |   |
| Calcium    | Ca     | 20            | 2                                         | 8 | 8 | 2 |   |   |

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●  $c = v\lambda$

●  $\bar{v} = \frac{1}{\lambda} = \frac{v}{c}$

[ $c$  = velocity of light in free space,

$v$  = frequency,  $\lambda$  = wavelength]

[ $\bar{v}$  = wave number]

# Physical Science Notes Chapter 4 Atomic Structure

**Wolfgang Demtröder**



## **Physical Science Notes Chapter 4 Atomic Structure:**

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Physical Science Notes Chapter 4 Atomic Structure**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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