

- a) Benzene.
- b) Octane.
- c) Pentane.
- d) Entahol.

**ANSWER : Pentane.**

**39) When a solute is dissolved in water it shows:**

- a) decrease in freezing point of water.
- b) decrease in boiling point of water.
- c) increase in vapour pressure of water.
- d) all of the above.

**ANSWER : decrease in freezing point of water.**

**40) When sugar is added to water, what is the change observed in boiling and freezing points of water?**

- a) Both boiling point and freezing point decreases.
- b) Both boiling point and freezing point increases.
- c) Boiling point increases and freezing point decreases.
- d) Boiling point decreases and freezing point increases.

**ANSWER : Boiling point increases and freezing point decreases.**

**41) ..... aqueous molal solution have highest freezing point.**

- a) Urea
- b) Barium chloride.
- c) Potassium bromide.
- d) Aluminium sulphate.

**ANSWER : Urea**

**42) When a thin slice of sugar beet is placed in a concentrated solution of NaCl then sugar beet will.....**

- a) dissolve in solution.
- b) neither absorb nor lose water.
- c) absorb water from solution.
- d) lose water from its cell.

**ANSWER : lose water from its cell.**

**43) The wrong relation between osmotic pressure (P), volume (V) and temperature (T) is:**

- a)  $P \propto n$  if T and V are constant.
- b)  $Pv$  is constant if T is constant.
- c)  $P \propto V$  if T is constant.
- d)  $P \propto T$  if V is constant.

**ANSWER :  $P \propto V$  if T is constant.**

**44) Two solutions C and D are separated by a semi-permeable membrane. If liquid flows from D to C then.**

- a) Both have same concentration.
- b) D is less concentrated than C.
- c) D is more concentrated than C.
- d) None of these.

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**Lingjun Ying**



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