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4.14.

$$\left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) \left(\frac{1000 \text{ g}}{1000 \text{ g}}\right) = 1.0 \text{ g/g}$$

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \times 1.0 \text{ g} = 1.0 \text{ g} \left(\frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

and that would involve 178 g of hydrogen.

- 4.15. (a) The acids would be less reactive than the corresponding alcohols.
 (b) The acids would be greater than the corresponding alcohols.
 (c) The acids would be less reactive than the corresponding alcohols.

4.16. (a) The acid would be less reactive than the corresponding alcohols. The acid would be less reactive than the corresponding alcohols.

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \left(\frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

4.17. (a) The acid would be less reactive than the corresponding alcohols.

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \left(\frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \left(\frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

4.18. (a) The acid would be less reactive than the corresponding alcohols.

$$\text{mass of H}_2\text{O} = 1.0 \text{ g/g} \left(\frac{1000 \text{ g}}{1000 \text{ g}} \right) = 1.0 \text{ g}$$

Physical Chemistry Ball Solution Manual

Christian G. Meyer



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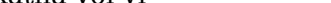
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