

ALKALI METALS

ALKALINE EARTH METALS

TRANSITION METALS

LANTHANIDES AND ACTINIDES

The image shows a standard periodic table of elements, color-coded by groups. Three callout boxes highlight specific categories:

- ALKALINE EARTH METALS:** A blue box points to the second column (Group 2), which includes elements like Lithium (Li), Beryllium (Be), Sodium (Na), Magnesium (Mg), Potassium (K), Calcium (Ca), Rubidium (Rb), Strontium (Sr), Cesium (Cs), Barium (Ba), Francium (Fr), and Radium (Ra).
- TRANSITION METALS:** A yellow box points to the d-block, which includes elements from Groups 3 to 10, such as Scandium (Sc) through Nickel (Ni) in the first row, and Lanthanum (La) through Platinum (Pt) in the second row.
- LANTHANIDES AND ACTINIDES:** A light blue box points to the two rows of elements at the bottom of the table, which are the Lanthanide and Actinide series, including elements from Lanthanum (La) to Lutetium (Lu) and Actinium (Ac) to Lawrencium (Lr).

The periodic table is organized into rows (periods) and columns (groups). The groups are labeled with Roman numerals at the top: I, II, III, IV, V, VI, VII, VIII, IX, X, XI, XII, XIII, XIV, XV, XVI, XVII, XVIII. The elements are color-coded by group: Group 1 (pink), Group 2 (blue), Groups 3-10 (yellow), Groups 11-12 (orange), Groups 13-16 (green), Group 17 (red), Group 18 (light blue), and the Lanthanides/Actinides (light blue).

Periodic Table Physical Properties Study

Disha Experts



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