

Name: \_\_\_\_\_

**Dates:**

## CHEMISTRY

## Periodic Table of the Elements

Use the Periodic Table of the Elements to answer the questions.

The image shows a standard periodic table of elements. The elements are color-coded by groups:
 

- Group 1 (Alkali metals): Light blue
- Group 2 (Alkaline earth metals): Orange
- Groups 3-10 (Transition metals): Yellow
- Group 11 (Coinage metals): Light green
- Group 12 (Zn, Cd, Hg): Dark green
- Group 13 (Boron group): Light blue
- Group 14 (Carbon group): Blue
- Group 15 (Nitrogen group): Purple
- Group 16 (Chalcogens): Pink
- Group 17 (Halogens): Red
- Group 18 (Noble gases): Light yellow
- Period 7 elements (Fr, Ra, Ac-Lr): Dark blue
- Lanthanides and Actinides (bottom rows): Dark blue

Labels for Boron (B) and its isotopes are present:
 

- B**: Element symbol for Boron, located in Group 13, Period 2.
- Boron**: Element name, located below the symbol B.
- Atomic number**: 5, located to the left of the symbol B.
- Atomic mass**: 10.811, located to the right of the symbol B.
- Isotopes**: A label pointing to the isotopes section, located to the right of the symbol B.
- Group**: A label pointing to the group number, located to the right of the symbol B.
- Period**: A label pointing to the period number, located to the right of the symbol B.

Isotopes of Boron are listed in the top right corner:
 

- B-10**: Stable isotope, 10.0129370 u, 19.9% abundance.
- B-11**: Stable isotope, 11.0093055 u, 80.1% abundance.
- B-12**: Radioactive isotope, 12.0143541 u, 0.0% abundance.
- B-13**: Radioactive isotope, 13.0033548 u, 0.0% abundance.
- B-14**: Radioactive isotope, 14.0030740 u, 0.0% abundance.
- B-15**: Radioactive isotope, 15.0030686 u, 0.0% abundance.
- B-16**: Radioactive isotope, 16.0030740 u, 0.0% abundance.
- B-17**: Radioactive isotope, 17.0030740 u, 0.0% abundance.
- B-18**: Radioactive isotope, 18.0030740 u, 0.0% abundance.
- B-19**: Radioactive isotope, 19.0030740 u, 0.0% abundance.
- B-20**: Radioactive isotope, 20.0030740 u, 0.0% abundance.
- B-21**: Radioactive isotope, 21.0030740 u, 0.0% abundance.
- B-22**: Radioactive isotope, 22.0030740 u, 0.0% abundance.
- B-23**: Radioactive isotope, 23.0030740 u, 0.0% abundance.
- B-24**: Radioactive isotope, 24.0030740 u, 0.0% abundance.
- B-25**: Radioactive isotope, 25.0030740 u, 0.0% abundance.
- B-26**: Radioactive isotope, 26.0030740 u, 0.0% abundance.
- B-27**: Radioactive isotope, 27.0030740 u, 0.0% abundance.
- B-28**: Radioactive isotope, 28.0030740 u, 0.0% abundance.
- B-29**: Radioactive isotope, 29.0030740 u, 0.0% abundance.
- B-30**: Radioactive isotope, 30.0030740 u, 0.0% abundance.
- B-31**: Radioactive isotope, 31.0030740 u, 0.0% abundance.
- B-32**: Radioactive isotope, 32.0030740 u, 0.0% abundance.
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- B-37**: Radioactive isotope, 37.0030740 u, 0.0% abundance.
- B-38**: Radioactive isotope, 38.0030740 u, 0.0% abundance.
- B-39**: Radioactive isotope, 39.0030740 u, 0.0% abundance.
- B-40**: Radioactive isotope, 40.0030740 u, 0.0% abundance.
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- B-86**: Radioactive isotope, 86.0030740 u, 0.0% abundance.
- B-87**: Radioactive isotope, 87.0030740 u, 0.0% abundance.
- B-88**: Radioactive isotope, 88.0030740 u, 0.0% abundance.
- B-89**: Radioactive isotope, 89.0030740 u, 0.0% abundance.
- B-90**: Radioactive isotope, 90.0030740 u, 0.0% abundance.
- B-91**: Radioactive isotope, 91.0030740 u, 0.0% abundance.
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- B-98**: Radioactive isotope, 98.0030740 u, 0.0% abundance.
- B-99**: Radioactive isotope, 99.0030740 u, 0.0% abundance.
- B-100**: Radioactive isotope, 100.0030740 u, 0.0% abundance.
- B-101**: Radioactive isotope, 101.0030740 u, 0.0% abundance.
- B-102**: Radioactive isotope, 102.0030740 u, 0.0% abundance.
- B-103**: Radioactive isotope, 103.0030740 u,

**Part 1: Atomic Number.** On the line next to the element, write the atomic number of the element. Your answer should be 1, 2, or 3 digits. No letters.

- |    |  |           |     |  |              |
|----|--|-----------|-----|--|--------------|
| 1. |  | Hydrogen  | 7.  |  | Cerium       |
| 2. |  | Oxygen    | 8.  |  | Iridium      |
| 3. |  | Magnesium | 9.  |  | Bismuth      |
| 4. |  | Vanadium  | 10. |  | Uranium      |
| 5. |  | Krypton   | 11. |  | Fermium      |
| 6. |  | Ruthenium | 12. |  | Darmstadtium |

# Periodic Table Answers Section 3

**G Thomas**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

### **Periodic Table Answers Section 3:**

## Unveiling the Power of Verbal Beauty: An Mental Sojourn through **Periodic Table Answers Section 3**

In a global inundated with displays and the cacophony of quick conversation, the profound power and emotional resonance of verbal beauty often fade in to obscurity, eclipsed by the regular onslaught of sound and distractions. Yet, set within the lyrical pages of **Periodic Table Answers Section 3**, a charming work of fictional elegance that impulses with raw emotions, lies an remarkable trip waiting to be embarked upon. Written with a virtuoso wordsmith, that mesmerizing opus guides visitors on a mental odyssey, gently revealing the latent possible and profound influence stuck within the delicate internet of language. Within the heart-wrenching expanse with this evocative examination, we can embark upon an introspective exploration of the book is key subjects, dissect its captivating publishing design, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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### **Table of Contents Periodic Table Answers Section 3**

1. Understanding the eBook Periodic Table Answers Section 3
  - The Rise of Digital Reading Periodic Table Answers Section 3
  - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Answers Section 3
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Periodic Table Answers Section 3
  - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Answers Section 3
  - Personalized Recommendations

- Periodic Table Answers Section 3 User Reviews and Ratings
- Periodic Table Answers Section 3 and Bestseller Lists
- 5. Accessing Periodic Table Answers Section 3 Free and Paid eBooks
  - Periodic Table Answers Section 3 Public Domain eBooks
  - Periodic Table Answers Section 3 eBook Subscription Services
  - Periodic Table Answers Section 3 Budget-Friendly Options
- 6. Navigating Periodic Table Answers Section 3 eBook Formats
  - ePub, PDF, MOBI, and More
  - Periodic Table Answers Section 3 Compatibility with Devices
  - Periodic Table Answers Section 3 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Periodic Table Answers Section 3
  - Highlighting and Note-Taking Periodic Table Answers Section 3
  - Interactive Elements Periodic Table Answers Section 3
- 8. Staying Engaged with Periodic Table Answers Section 3
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Periodic Table Answers Section 3
- 9. Balancing eBooks and Physical Books Periodic Table Answers Section 3
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Periodic Table Answers Section 3
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Answers Section 3
  - Setting Reading Goals Periodic Table Answers Section 3
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Answers Section 3
  - Fact-Checking eBook Content of Periodic Table Answers Section 3

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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