

Seat No.	
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T.Y. B.Sc. (Semester – III) Examination, 2012
CHEMISTRY (Paper – III)
CH – 333 : Organic Chemistry (2008 Pattern)

Time : 2 Hours

Max. Marks : 40

- N.B. :** i) All questions are **compulsory**.
 ii) Figures to the **right** indicate **full marks**.
 iii) **Draw** the structures and **neat** diagram if necessary.

1. Answer the following :

10

- i) Why urea is neutral though it contains —NH_2 group ?
- ii) Write the trivial and IUPAC name for ph—CH=CH—C(=O)—ph .
- iii) Draw Zig-Zag structure of hepta, 3-5-dione.
- iv) Arrange the following in decreasing order of $\text{S}_{\text{N}}2$ reactivity. Explain your reasoning.



- v) Guanidine is extremely strong base-explain in brief.
- vi) 2-phenyl ethyl bromide on heating with NaOCH_3 gives 95 % styrene.
- vii) Write the reaction of Ethyl thiol with benzaldehyde.
- viii) 1-Butene treated with HBr in presence of peroxide. Write the product.
- ix) What is $\text{S}_{\text{N}}1$ reaction ?
- x) Salicylic acid is stronger acid than m-hydroxy benzoic acid - why ?

P.T.O.

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Table of Contents Organic Chemistry Question Papers University

1. Understanding the eBook Organic Chemistry Question Papers University
 - The Rise of Digital Reading Organic Chemistry Question Papers University
 - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chemistry Question Papers University
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Organic Chemistry Question Papers University
 - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chemistry Question Papers University
 - Personalized Recommendations
 - Organic Chemistry Question Papers University User Reviews and Ratings
 - Organic Chemistry Question Papers University and Bestseller Lists
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 - Highlighting and Note-Taking Organic Chemistry Question Papers University
 - Interactive Elements Organic Chemistry Question Papers University
- 8. Staying Engaged with Organic Chemistry Question Papers University
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Organic Chemistry Question Papers University
- 9. Balancing eBooks and Physical Books Organic Chemistry Question Papers University
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Organic Chemistry Question Papers University
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chemistry Question Papers University
 - Setting Reading Goals Organic Chemistry Question Papers University
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chemistry Question Papers University
 - Fact-Checking eBook Content of Organic Chemistry Question Papers University
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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