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Chemistry Practice Worksheet: Chemical Formulas

Write a chemical formula for each compound:

1. Al_2O_3 aluminum oxide
2. Hg_2Cl_2 mercurous chloride
3. H_2SO_4 sulfuric acid
4. K_2SO_4 potassium sulfate
5. CH_3COOH acetic acid
6. $AgNO_3$ silver nitrate
7. $Ca(OH)_2$ calcium hydroxide
8. Na_2CO_3 sodium carbonate
9. $CaCl_2$ calcium chloride
10. Na_2SO_4 sodium sulfate
11. Fe_2O_3 iron(III) oxide
12. Na_2O sodium oxide
13. $Ca_3(PO_4)_2$ calcium phosphate
14. CH_4 methane
15. CO_2 carbon dioxide
16. H_2O water
17. $NaCl$ sodium chloride
18. CaF_2 calcium fluoride
19. Al_2S_3 aluminum sulfide
20. Fe_2S_3 iron(III) sulfide
21. Na_2S sodium sulfide
22. CaS calcium sulfide
23. Al_2O_3 aluminum oxide
24. Na_2O sodium oxide
25. CaO calcium oxide
26. K_2O potassium oxide
27. FeO iron(II) oxide
28. CO carbon monoxide
29. H_2CO_3 carbonic acid
30. $CaCO_3$ calcium carbonate
31. $NaHCO_3$ sodium bicarbonate
32. Na_2CO_3 sodium carbonate
33. Na_2SO_4 sodium sulfate
34. Na_2CO_3 sodium carbonate
35. Na_2CO_3 sodium carbonate
36. Na_2CO_3 sodium carbonate
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47. Na_2CO_3 sodium carbonate
48. Na_2CO_3 sodium carbonate
49. Na_2CO_3 sodium carbonate
50. Na_2CO_3 sodium carbonate

51. $CaCl_2$ calcium chloride
52. $NaCl$ sodium chloride
53. Na_2SO_4 sodium sulfate
54. Na_2CO_3 sodium carbonate
55. Na_2O sodium oxide
56. $CaCl_2$ calcium chloride
57. Na_2O sodium oxide
58. Na_2CO_3 sodium carbonate
59. Na_2CO_3 sodium carbonate
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