Grade 8 Science Notes: Chemical Reactions

Chemical Reactions are processes where substances are transformed into new substances with different properties.

Types of Chemical Reactions:

- 1. Combination Reaction:
- Two or more substances combine to form one product.
- Example: Hydrogen + Oxygen -> Water.
- 2. Decomposition Reaction:
 - A compound breaks down into simpler substances.
 - Example: Water -> Hydrogen + Oxygen.
- 3. Displacement Reaction:
 - An element replaces another in a compound.
 - Example: Zinc + Copper Sulfate -> Zinc Sulfate + Copper.
- 4. Combustion Reaction:
 - A substance reacts with oxygen, releasing energy.
 - Example: Methane + Oxygen -> Carbon Dioxide + Water + Energy.

Reactants and Products:

- Reactants: Starting substances.
- Products: New substances formed.

Indicators of a Chemical Reaction:

- 1. Color Change.
- 2. Gas Production (bubbles).
- 3. Temperature Change (heat release/absorption).
- 4. Formation of Precipitate.

Importance:

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Chemical News and Journal of Industrial Science ,1882 The Chemical News and Journal of Industrial **Science** ,1882 English Mechanic and Mirror of Science and Art ,1883 University of Michigan Official Publication University of Michigan, 1988 Each number is the catalogue of a specific school or college of the University English Mechanic and World of Science ,1883 Simulation Algorithms for Computational Systems Biology Luca Marchetti, Corrado Priami, Vo Hong Thanh, 2017-09-27 This book explains the state of the art algorithms used to simulate biological dynamics Each technique is theoretically introduced and applied to a set of modeling cases Starting from basic simulation algorithms the book also introduces more advanced techniques that support delays diffusion in space or that are based on hybrid simulation strategies This is a valuable self contained resource for graduate students and practitioners in computer science biology and bioinformatics An appendix covers the mathematical background and the authors include further reading sections in each chapter Nuclear Science Abstracts ,1972 Hard and Soft Acids and Bases Ralph G. Pearson, 1973

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