

Physical Science  
Chapter 2

## Properties of Matter

### Classifying Matter

#### Substances

- **Element:** a substance that can't be separated down into simpler substances.
- **Atom:** the smallest particle of an element.
- An element has a fixed composition because it contains only one type of atom.

#### 2 Types of Mixtures

1. **Heterogeneous Mixture:** have materials that are not evenly distributed and noticeably different from one another.
  2. **Homogeneous Mixture:** have materials that are evenly distributed so it is difficult to distinguish one substance from another.
    - a. Also called a solution
    - b. Remain constantly and uniformly mixed
- Which type of mixture is milk?
    - Based on the size of its largest particles, a mixture can be classified as a solution, a suspension, or a colloid.

#### Solutions

**Solution:** when substances dissolve and form a homogenous mixture.

- Examples: windshield wiper fluid, sea water, lemonade

#### Colloids

**Colloid:** contains some particles that are larger than those in a solution, but smaller than those in a suspension.

- Examples: homogenized milk, fog
- Think about it...
  - What happens to the size of the drops of fat in milk when milk is homogenized?

#### The Scattering of Light

- Colloids and Suspensions scatter light just like fog.
- Light goes right through a solution.

#### Chemical Properties

- Any ability to produce a change in the composition of matter
- Chemical properties can be observed only when the substance in a sample of matter is changing into different substances.

# Physical Science Chapter 2 Review Answers Properties Of Matter

**Walter George Whitman**



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