

PERIODIC TABLE

Atomic Properties of the Elements

NIST
National Institute of
Standards and Technology
U.S. Department of Commerce

Frequently used fundamental physical constants																		Physics Laboratory		Standard Reference Data		2	
For the most accurate values of these and other constants, visit physics.nist.gov/constants																		physics.nist.gov		www.nist.gov/srd			
1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³ Cs																							
speed of light in vacuum c 299 792 458 m s ⁻¹ (exact)																							
Planck constant h 6.626 070 15 × 10 ⁻³⁴ J s (exact)																							
elementary charge e 1.602 176 634 × 10 ⁻¹⁹ C																							
electron mass m _e 9.109 383 56 × 10 ⁻³¹ kg																							
proton mass m _p 1.672 621 9 × 10 ⁻²⁷ kg																							
fine-structure constant α 1/137.036																							
Rydberg constant R _∞ 10 973 731.57 m ⁻¹																							
Bohrmann constant k _B 1.380 658 × 10 ⁻²³ J K ⁻¹																							
Boltzmann constant k 1.380 658 × 10 ⁻²³ J K ⁻¹																							

Atomic Number: 58
Ground-state Ionization Energy (eV): 5.5337

Symbol: **Ce**
Name: Cerium
Atomic Weight: 140.116
Ground-state Configuration: [Xe]4f¹5d¹6s²

*Based upon ¹²C. () indicates the mass number of the longest-lived isotope.

For a description of the data, visit physics.nist.gov/data

Periodix Table Study Packet

CH Cherryholmes



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