

## Chemical Periodicity

Name Mitchell Wade Section Number 015

**OBJECTIVE:** To demonstrate the periodic law.

**CONCEPTS:** Periodic law, Periodic trends

**INTRODUCTION:** In the 1860's about 60 elements were known. From their physical properties, the elements were classified as metals, nonmetals, or metalloids. Metals have low electrical resistance (high conductivity), have a metallic luster, and are malleable. Non-metals have high resistance, low luster, and are brittle. Metalloids have properties between those of non-metals and metals. Chemists of the time further classified the elements into groups. The elements in a particular group have similar chemical and/or physical properties. This classification led to the discovery of the periodic table and the periodic law. In the periodic table, elements are assigned to a period and a group—groups are the columns and the periods are the rows of the table. The periodic law states that the properties of the chemical elements vary with the atomic number in a systematic way. We now understand that the properties depend on the electronic structures of the atoms.

### PROCEDURE:

1. Find the station labeled "Halogens: states of matter." Observe the states of matter of the three samples. How does the state of matter change as you go down the column of the periodic table?

Starts as gas and goes to solid

The melting points of the halogen elements that you observed are listed below. Based on the trend that you see, estimate the melting point of Fluorine.

Element	Melting Point ( $^{\circ}\text{C}$ )
F	-200
Cl	-101.5
Br	-7.35
I	113.75

2. Find the station labeled "Metallic Solutions." Fill out the chart below. For the unlabeled mystery elements, you will have to guess which type of metal they are (transition, alkali, or alkaline earth) based on the trends you see.

Solution Number	Element	Color of Water Solution	Type (transition metal, alkali metal, alkaline earth metal)
1	Co	Red	Transition metal
2	Ba	Clear	Alkaline earth metal
3	K	clear	Alkali metal
4	Ni	green	Transition metal
5	Na	clear	Alkali metal
6	Cu	blue	Transition metal
7	Mg	clear	Alkaline earth metal

# Periodicity Lab Report

**JS Bruner**



## **Periodicity Lab Report:**

This book delves into Periodicity Lab Report. Periodicity Lab Report is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Periodicity Lab Report, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Periodicity Lab Report
- Chapter 2: Essential Elements of Periodicity Lab Report
- Chapter 3: Periodicity Lab Report in Everyday Life
- Chapter 4: Periodicity Lab Report in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Periodicity Lab Report. This chapter will explore what Periodicity Lab Report is, why Periodicity Lab Report is vital, and how to effectively learn about Periodicity Lab Report.
3. In chapter 2, the author will delve into the foundational concepts of Periodicity Lab Report. The second chapter will elucidate the essential principles that need to be understood to grasp Periodicity Lab Report in its entirety.
4. In chapter 3, this book will examine the practical applications of Periodicity Lab Report in daily life. The third chapter will showcase real-world examples of how Periodicity Lab Report can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Periodicity Lab Report in specific contexts. This chapter will explore how Periodicity Lab Report is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Periodicity Lab Report. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Periodicity Lab Report.

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