

Light & Periodicity HOMEWORK

Abstract

1. **Wagner's first concern** was to ensure that the 1934 amendment to the Copyright Act was passed.

- | | |
|----|---------------|
| Q1 | 12 months ago |
| Q2 | 9 months ago |
| Q3 | 6 months ago |
| Q4 | 3 months ago |
| Q5 | 12 months ago |

3. **RESEARCH DESIGN** The study was a quasi-experimental design.

1. **Introduction**
 2. **Background**
 3. **Methodology**
 4. **Results**
 5. **Conclusion**

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14. An object is moving through water, pulled against the resistance. There is a 2000 N force opposing the motion.

10. A signal, which is a sinusoidal wave, travels a mean free distance (a wavelength) of 1.00×10^{-2} meters. What is its frequency?

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[illegible]

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5. Light can be used to excite the electrons in atoms. Certain frequencies of light work best for specific atoms. To excite an atom of Calcium, the ideal frequency of light is 3.2×10^{14} Hz. What is the wavelength (in nm) of this electromagnetic radiation? Give your answer in scientific notation and with units.

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8. The energy required to dissociate the Cl_2 molecule into two separate Cl atoms requires a wavelength of 5.00×10^{-7} meters. What is the energy of this light in joules?

† *Indicated as the ^{100}Mg isotope is 100% in ^{100}Mg primary standard (NIST SRM 984) as the ^{100}Mg isotope is 100% in ^{100}Mg standard.*

$\lim_{t \rightarrow \infty} \frac{1}{t} \log \frac{1}{\mathbb{P}(\tau \leq t)} = -\lambda$
 $\lim_{t \rightarrow \infty} \frac{1}{t} \log \frac{1}{\mathbb{P}(\tau \leq t)} = -\lambda$
 $\lim_{t \rightarrow \infty} \frac{1}{t} \log \frac{1}{\mathbb{P}(\tau \leq t)} = -\lambda$

1. Refrigerator coolant contains organic compounds known as chlorofluorocarbons (CFCs). If not properly disposed, a refrigerator can leak this gas into the environment. There, ultraviolet radiation from the sun can change these CFCs, which can then destroy the ozone layer. How much energy do the CFCs receive from UV light that has a frequency $4.15 \times 10^{14} \text{ s}^{-1}$? (Give your answer in joules.)

$\text{Dose}_{\text{max}} = 100 \text{ mg}$
 $\text{Dose}_{\text{min}} = 10 \text{ mg}$
 $\text{Dose}_{\text{max}} = 100 \text{ mg}$

15. You want your new speakers to come on in the morning, at first light. So, you design a switch that will operate by the photoelectric effect when enough sunlight hits it. As sunlight hits on your switch, the metal inside the device absorbs energy and ejects electrons to create electricity that powers the speakers. If the switch requires 3.8×10^{-19} joules of energy to remove an electron, will the switch work if the sunlight has a wavelength of 540 nm?

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