

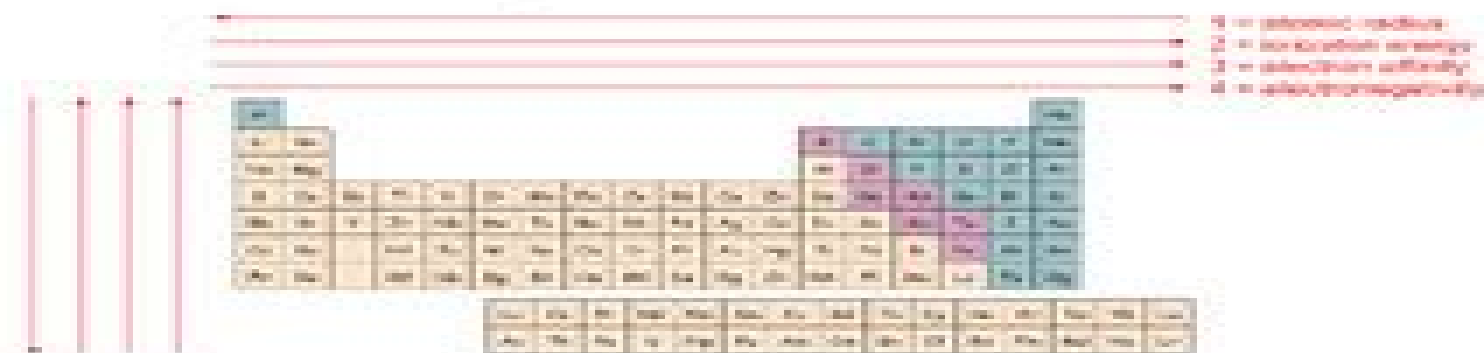
## Periodic Trends - Mixed Review

## Answer Key

1. Match each definition with the correct trend below.

- |                      |  |
|----------------------|--|
| d. Ionization Energy | a. The energy change when one mole of electrons is added to one mole of gaseous atoms, forming negatively charged ions.                |
| c. Atomic Radius     | b. A measure of an atom's ability to attract electrons in a chemical bond when forming a compound.                                     |
| b. Electronegativity | c. The distance from the nucleus to the outermost electron in an atom, or half the distance between two nuclei in a diatomic molecule. |
| a. Electron Affinity | d. The energy required to remove one mole of electrons from one mole of gaseous atoms.   |

2. Draw and label arrows on the periodic table below that indicate the direction of each trend.



3. Arrange the following elements in order of increasing atomic radius:

Be, Cl, F, I

F, Cl, Be, I

4. Rank the following elements in order of decreasing atomic radius:

Cl, Li, B, Be

Li, Be, B, Cl

5. Place the following elements in order of increasing ionization energy:

Li, K, F, O

K, Li, O, F

# Periodic Trends And Mixed Review

**Wenbin Ji**

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red circular glow behind it.

## **Periodic Trends And Mixed Review:**

The book delves into Periodic Trends And Mixed Review. Periodic Trends And Mixed Review is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Periodic Trends And Mixed Review, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
  - Chapter 1: Introduction to Periodic Trends And Mixed Review
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  - Chapter 3: Periodic Trends And Mixed Review in Everyday Life
  - Chapter 4: Periodic Trends And Mixed Review in Specific Contexts
  - Chapter 5: Conclusion
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