

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

### Periodic Trends Practice Worksheet

1. What is the difference between electronegativity and ionization energy?
2. Why does fluorine have a higher ionization energy than iodine?
3. Why do elements in the same family generally have similar properties?
4. Which is the largest atom in Group 4a (14)?
5. Which is the smallest atom in Group 7a (17)?
6. Which is the smallest atom in period 5?
7. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.
8. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.
9. Arrange the following atoms in order of decreasing atomic radius. Na Al P Cl Mg
10. For each of the following pairs, circle the element that is larger?  
a)  $\text{N}^{3-}$  or  $\text{F}^{-}$       b)  $\text{Mg}^{2+}$  or  $\text{Ca}^{2+}$       c)  $\text{Fe}^{2+}$  or  $\text{Fe}^{3+}$
11. Circle the element in each pair has the larger radius?  
a) Mg or  $\text{Mg}^{2+}$       b) O or  $\text{O}^{2-}$       c)  $\text{K}^{+}$  or  $\text{Cl}^{-}$       d)  $\text{P}^{3-}$  or  $\text{S}^{2-}$
12. In each of the following pairs, circle the species with the higher first ionization energy:  
a) Li or Cs      (b) Cl or Ar      (c) Ca or Br      (d) Na or Ne      (e) B or Be
13. In each of the following pairs, circle the species with the larger atomic radius:  
a) Mg or Ba      (b) S or  $\text{S}^{2-}$       (c)  $\text{Cu}^{+2}$  or Cu      (d) He or H      (e) Na or Cl
14. Circle the best choice in each list:  
a) highest first ionization energy: C, N, Si  
b) highest electronegativity: As, Sn, S  
c) largest radius:  $\text{S}^{2-}$ ,  $\text{Cl}^{-}$ , Cl
15. Order the following groups from largest to smallest radii.  
a) Ar, Cl, K,  $\text{S}^{2-}$   
b) C, Al, F, Si  
c) Na, Mg, Ar, P  
d)  $\text{I}^{-}$ ,  $\text{Ba}^{2+}$ ,  $\text{Cs}^{+1}$ ,  $\text{F}^{-1}$

# Periodic Trends Extension Questions Worksheet

**Michael Brown**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Periodic Trends Extension Questions Worksheet:**

## Unveiling the Energy of Verbal Artistry: An Emotional Sojourn through **Periodic Trends Extensoin Questions Worksheet**

In a global inundated with displays and the cacophony of fast connection, the profound power and mental resonance of verbal art often disappear in to obscurity, eclipsed by the continuous barrage of noise and distractions. Yet, located within the musical pages of **Periodic Trends Extensoin Questions Worksheet**, a interesting function of fictional splendor that impulses with fresh feelings, lies an unforgettable trip waiting to be embarked upon. Penned with a virtuoso wordsmith, that mesmerizing opus courses viewers on a psychological odyssey, delicately exposing the latent possible and profound influence stuck within the complicated internet of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is key themes, dissect their fascinating writing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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