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THE PERIODIC TABLE

Chapter 6 Review Package - Part 1

A. Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

	Column A		Column B
33	metal	100	positively charged ion
2.	halogen	b.	an element whose highest occupied s or ρ sublevels are partially filled
3.	inner transition metal	c.	the energy needed to remove an electron from an atom in the gaseous state
4.	representative element	d.	good conductor of heat and electric current
5	ionization energy	e.	an element whose highest occupied s sublevel and a nearby d sublevel contain electrons
6.	alkaline earth metal	ď.	any nonmetal in Group 7A
7.	cation	g.	the tendency of an atom to attract electrons when the atom is in a compound
8.	electronegativity	h.	an element in which the highest occupied s and ρ sublevels are filled
9.	transition metal	1.0	an element in Group 2A
10.	noble gas	J.	an element whose highest occupied s sublevel and a nearby f sublevel contain electrons

- 11. In the periodic table, there is a periodic pattern in the physical and chemical properties of elements when they are arranged in order of a. increasing atomic mass.
 - b. increasing electronegativity.
 - e. increasing atomic radius.

 - d. increasing atomic number.



Periodic Trends Ch 6 Answers

ML Yell

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