

Step 8. Periodic table trends

Electronegativity

Ionization energy

He

Atomic radius

The periodic table displays various elements with their symbols and atomic numbers. Overlaid on the table are several arrows indicating trends:

- Electronegativity:** A black arrow pointing from left to right across the top of the table.
- Ionization energy:** A green arrow pointing from left to right across the top of the table, and a green arrow pointing from bottom to top on the right side of the table.
- Atomic radius:** A red arrow pointing from top to bottom on the left side of the table, and a red arrow pointing from right to left across the bottom of the table.
- Non-metallic properties:** A purple arrow pointing from the bottom-left towards the top-right.
- Metallic properties:** A blue arrow pointing from the top-left towards the bottom-right.

Ionization Energy

Electronegativity

Atomic radius

An arrow head shows the INCREASE of the property

Example: comparing K atom with Br atom

- Atomic radius of K is HIGHER than atomic radius of Br
- Ionization energy of K is LOWER than ionisation energy of Br
- Metallic properties of K are HIGHER than metallic properties of Br



Periodic Trends Answer Key Section Review

Tao Wei



Periodic Trends Answer Key Section Review:

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In a digitally-driven earth where screens reign supreme and quick transmission drowns out the subtleties of language, the profound techniques and psychological nuances hidden within words usually get unheard. Yet, situated within the pages of **Periodic Trends Answer Key Section Review** a charming literary prize blinking with organic thoughts, lies a fantastic quest waiting to be undertaken. Penned by a talented wordsmith, this marvelous opus invites readers on an introspective journey, gently unraveling the veiled truths and profound influence resonating within ab muscles material of every word. Within the psychological depths of this moving evaluation, we can embark upon a heartfelt exploration of the book is core subjects, dissect their fascinating publishing type, and yield to the effective resonance it evokes strong within the recesses of readers hearts.

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