

Name: _____

Atomic Structure & the Periodic Table Study Guide

Atomic Structure

| Particle | Charge | Where are they found? |
|----------|--------------------|--|
| Proton | Positive | Nucleus |
| Neutron | Neutral/ no charge | Nucleus |
| Electron | negative | Electron cloud/energy levels outside nucleus |

1. What is the rule for writing chemical symbols like Na or Ca? **Capitalize the first letter; lower case second letter**
2. What are the vertical columns on the periodic table called? Groups or families
 - a. How many of these are on the periodic table? 18
 - b. Why are the elements in these columns? They have the same chemical properties (& same # of valence electrons)
 - c. Elements in the same group have the same number of valence electrons.
3. What are the horizontal rows on the periodic table called? periods
 - a. How many of these are on the periodic table? 7
 - b. Elements in these rows have the same number of energy (*electron shells*) levels.
4. What are valence electrons? **Electrons in the outer energy level**
5. How many electrons are in each energy level?
1st energy level 2 2nd energy level 8 3rd energy level 8
6. The atomic number is equal to the # of protons and also # of electrons
7. The atomic mass is equal to the # protons + # of neutrons
8. How do you find the number of neutrons? Atomic mass – atomic number
9. We view the order of the periodic table today according to the atomic number

Periodic Table Printable Study Guide

Julia Schneider



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interview questions and answers on Chemical bonding chemical combinations atomic radii atomic radius periodic table atomic ionic and covalent radii atoms and molecules bond formation covalent radius electron affinity electronegativity electronegativity periodic table higher ionization energies ionic radius ionization energies ionization energy periodic table Lewis concept and modern periodic table The Experimental Techniques Quiz Questions PDF e Book Chapter 4 interview questions and answers on Experimental techniques chromatography crystallization filter paper filtration filtration crucibles solvent extraction and sublimation The Gases Quiz Questions PDF e Book Chapter 5 interview questions and answers on Gas laws gas properties kinetic molecular theory of gases ideal gas constant ideal gas density liquefaction of gases absolute zero derivation applications of Daltons law Avogadro s law Boyle s law Charles law Daltons law diffusion and effusion Graham s law of diffusion ideality deviations kinetic interpretation of temperature liquids properties non ideal behavior of gases partial pressure calculations plasma state pressure units solid s properties states of matter thermometry scales and van der Waals equation The Liquids and Solids Quiz Questions PDF e Book Chapter 6 interview questions and answers on Liquid crystals types of solids classification of solids comparison in solids covalent solids properties of crystalline solids Avogadro number determination boiling point external pressure boiling points crystal lattice crystals and classification cubic close packing diamond structure dipole dipole forces dipole induced dipole forces dynamic equilibrium energy changes intermolecular attractions hexagonal close packing hydrogen bonding intermolecular forces London dispersion forces metallic crystals properties metallic solids metal s structure molecular solids phase changes energies properties of covalent crystals solid iodine structure unit cell and vapor pressure

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Standards-based Activities and Assessments for the Differentiated Classroom Carolyn Coil,2004 How to plan and implement differentiation using practical strategies teacher friendly directions and time saving techniques

Galaxy S 4 Survival Guide: Step-by-Step User Guide for Galaxy S 4: Getting Started, Using eMail, Taking Photos and Videos, and Learning Hidden Tips and Tricks Toly K, The Galaxy S 4 is one of the best selling Android phones on the market today As a result the Galaxy S 4 is packed full of new features never before seen on any Android phone such as air gestures multi window applications eye tracking and the ability to view a video while using another application This guide will introduce you to these new features and show you how to use

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The Inclusion Toolkit Sarah Herbert, 2011-08-10 With downloadable electronic resources Are you looking for ways to make your classroom more inclusive to children with learning difficulties The practical strategies suggested in this book are supported by current research into effective teaching and learning and the author shows you how inclusion friendly teaching could look in your classroom The book is aimed at the teacher who wishes to respond to the diversity of learning needs of children that are placed within their mainstream classroom without teaching each child differently The advice given for teaching all learners aged 7 to 14 will help you ensure that you are providing well for all your learners that only a very few will need substantially different support Rather than focussing on the differences between learners the common strategies that will enhance the learning of all based on the common concerns of class teachers working with pupils who have Special Educational Needs are identified Structured around the three areas of memory motivation and communication which are key to addressing inclusion friendly teaching the book covers the memory friendly classroom the motivation friendly classroom the communication friendly classroom Ideal for all classroom teachers working with children aged 7 to 14 and with plenty of supporting material available to download from the SAGE website which works in tandem with each chapter of the book this toolkit will make inclusive teaching a reality

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thus teaches about the basic phenomena in the physical world and the concepts developed to explain them shows that
science is a rational human endeavor with a long and continuing tradition involving many different cultures and people
develops facility in critical thinking reasoned argumentation evaluation of evidence mathematical modeling and ethical
values The treatment emphasizes not only what we know but also how we know it why we believe it and what effects that
knowledge has Why do we believe the Earth and planets revolve around the Sun Why do we believe that matter is made of
atoms How do relativity theory and quantum mechanics alter our conception of Nature and in what ways do they leave the
classical concepts unchanged What impact does the knowledge of finite energy resources have on our society How have
applications of fundamental science such as the steam engine the laser the electric generator the transistor affected our lives
How does the evidence for non scientific ideas such as UFOs ESP and the like differ from the evidence for accepted scientific
results **Chemistry** Rob Ritchie,2008-07 Divided into seven manageable day sections this timed revision programme
covers essential GCSE topics in double page spreads These spreads indicate how much time should be spent on each section

and combine clear and concise explanations flow charts spidergrams and illustrations with progress check questions and answers

The Extraordinary Chemistry of Ordinary Things, Study Guide Carl H. Snyder, David F. Dever, 1995-01-19
Examines the chemistry of the substances of our everyday world Our daily lives are immersed in chemicals an effective way to teach and learn chemistry is by examining the goods and substances that we use in our daily lives and that affect us and our environment

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