
The Periodic Law

Section Quiz: History of the Periodic Table

As Marjorie completed each question of this quiz, please check that her responses match statements or facts covered in her presentation.

1. In developing his periodic table, Mendeleev's observations were
 - a. accurate enough.
 - b. incorrect and significant.
 - c. lacking.
 - d. **incomplete.**
2. Mendeleev's periodic table did not include all elements he used to be showing elements were similar by physical properties such as
 - a. **properties.**
 - b. atomic numbers.
 - c. masses.
 - d. density.
3. Mendeleev predicted that the gaps in his periodic table represented
 - a. ions.
 - b. unknown elements.
 - c. unstable elements.
 - d. **undiscovered elements.**
4. The periodic table came before a periodic table based on increasing
 - a. atomic number was
 - b. lacking.
 - c. **Mendeleev.**
 - d. Moseley.
5. The periodic law describes the physical and chemical properties of
 - a. **elements.**
 - b. molecules.
 - c. ions.
 - d. compounds.

Periodic Law Section Quiz Answers

J Rink



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