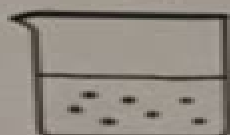


- Which illustration below represents
 - solute particles in a solid state in water?
 - solute particles in an aqueous state?



- What variables are controlled in all five beakers of Model 1?
 - amount of water - volume is the same
 - type of beakers
 - stirred for 2 hours
- Count the particles present in each beaker of Model 1. Fill in the table to show the number of dissolved solute particles and the number of solid solute particles.
- Consider the beakers in Model 1.
 - Which beakers represent unsaturated solutions?
B, D
 - Which beakers represent saturated solutions?
A, C, E
- Beakers A-E in Model 1 are depicted as representing five different or separate solutions. They could also be considered as five "snapshots" of the same beaker over time. In other words, if additional measured quantities of solute were stirred into beaker A in small increments over time, then beakers B-E would result.
 - When a small amount of additional solute is added to an unsaturated solution, what happens to the number of dissolved particles? Provide specific evidence from Model 1 to support your answer.
the number of dissolved particles increases
 - When a small amount of additional solute is added to a saturated solution, what happens to the number of dissolved particles? Provide specific evidence from Model 1 to support your answer.
the dissolved particles stay the same and the solid particles increase
 - Predict what would happen if a small amount of additional solute were stirred into beaker E in Model 1.
more solid particles would stay at the bottom



Pogil Solution Manual Equilibrium

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