

Model 2 – Proposed Calculations for Mass of NH_3 to Mass of N_2

Toby's Method

$$\frac{x \text{ grams}}{50.0 \text{ g}} = \frac{1 \text{ mole } \text{N}_2}{2 \text{ moles } \text{NH}_3} \rightarrow x = \underline{15} \text{ g } \text{N}_2$$

Rachel's Method

$$50.0 \text{ g } \text{NH}_3 \times \frac{1 \text{ mole } \text{NH}_3}{17.0 \text{ g } \text{NH}_3} = \underline{2.94} \text{ moles } \text{NH}_3$$

$$\frac{x \text{ mole } \text{N}_2}{\underline{1.5} \text{ mole } \text{NH}_3} = \frac{1 \text{ mole } \text{N}_2}{2 \text{ moles } \text{NH}_3} \rightarrow x = \underline{0.11} \text{ moles } \text{N}_2$$

$$\underline{0.11} \text{ mole } \text{N}_2 \times \frac{28.0 \text{ g } \text{N}_2}{1 \text{ mole } \text{N}_2} = \underline{2.43} \text{ g } \text{N}_2$$

Jerry's Method

$$50.0 \text{ g } \text{NH}_3 \times \frac{1 \text{ mole } \text{NH}_3}{17.0 \text{ g } \text{NH}_3} \times \frac{1 \text{ mole } \text{N}_2}{2 \text{ moles } \text{NH}_3} \times \frac{28.0 \text{ g } \text{N}_2}{1 \text{ mole } \text{N}_2} = \underline{2.43} \text{ g } \text{N}_2$$

- Model 2 shows three proposed calculations to solve the problem in Question 10. Complete the calculations in Model 2 by filling in the underlined values.
- Which method does not use the mole ratio in an appropriate manner? Explain.
Toby's method does not correctly use the mole ratio that is required.
- Two of the methods in Model 2 give the same answer. Show that they are mathematically equivalent methods.

$$\begin{aligned} &\text{Jerry's method} \\ &\left\{ \begin{aligned} &50 \times \frac{1}{17} = \frac{1}{2} \times \frac{28}{1} = 2.43 \text{ g } \text{N}_2 \\ &\quad \quad \quad \downarrow \\ &\quad \quad \quad \frac{1}{2} \times \frac{1}{2} \\ &\quad \quad \quad \downarrow \\ &0.11 \times \frac{28}{1} = 2.43 \text{ g } \text{N}_2 \end{aligned} \right. \end{aligned}$$

- Use either Rachel or Jerry's method from Model 2 to calculate the mass of hydrogen needed to make 50.0 g of ammonia. $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$

$$50.0 \text{ g} \left(\frac{1 \text{ mole}}{17.0 \text{ g}} \right) \left(\frac{3 \text{ moles}}{2 \text{ moles}} \right) \left(\frac{1 \text{ g}}{2.016 \text{ g}} \right) = 2.43 \text{ g } \text{H}_2 \text{ required}$$

Pogil Mole Ratios Model 2

**Thirumalaisamy P.
Velavancorresponding**



Pogil Mole Ratios Model 2:

Pogil Mole Ratios Model 2: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels enthralling the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the fascinating narratives that have captivated audiences this year. Pogil Mole Ratios Model 2 : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Pogil Mole Ratios Model 2 : Taylor Jenkins Reid's "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reid's captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Pogil Mole Ratios Model 2 : Delia Owens' "Where the Crawdads Sing" This mesmerizing coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is an exceptional and gripping novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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