

3. Refer to the data table in Question 2.

a. How do the reduced ratios in the last column compare to the coefficients in the reaction shown in Model 1?

The ratios are the same as the coefficients in Model 1: 1:3:2.

b. Use mathematical concepts to explain how your answer in part a is possible.

The coefficients were all multiplied by the same number, so they all ended back to the same thing.

4. Even 538 is a small number of molecules to use in a reaction. Typically chemists use much larger numbers of molecules. (Recall that one mole is equal to 6.02×10^{23} particles.) Consider each situation below as it relates to the reaction in Model 1: $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$.

a. Calculate the amount of reactants consumed and products made.

b. Record the ratio of N_2 to H_2 to NH_3 . Reduce the ratio to the lowest whole number possible.

	N_2 Consumed	H_2 Consumed	NH_3 Produced	Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$
If the reaction occurred 6.02×10^{23} times, how many molecules would be consumed or produced?	6.02×10^{23}	1.81×10^{24}	1.20×10^{24}	1 : 3 : 2
How many <i>moles</i> of each substance would be consumed or produced in the previous situation?	1	3	2	1 : 3 : 2

5. Refer to the data table in Question 4.

a. How do the reduced ratios in the last column compare to the coefficients in the reaction in Model 1?

The reduced ratios are the same as the coefficients.

b. Use mathematical concepts to explain how your answer in part a is possible.

We are still multiplying coefficients by the same number so we just didn't everything by the number to get back to a reduced ratio.

6. The ratio obtained from the coefficients in a balanced chemical equation is called the **mole ratio**.

a. What is the mole ratio for the reaction in Model 1?



b. Explain why this ratio is called the mole ratio?

It is the reduced ratio in moles (all the coefficients = 6.02×10^{23}).

Pogil Mole Ratio Key

Michael Rajnik

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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Table of Contents Pogil Mole Ratio Key

1. Understanding the eBook Pogil Mole Ratio Key
 - The Rise of Digital Reading Pogil Mole Ratio Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Mole Ratio Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Mole Ratio Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Mole Ratio Key
 - Personalized Recommendations
 - Pogil Mole Ratio Key User Reviews and Ratings
 - Pogil Mole Ratio Key and Bestseller Lists
5. Accessing Pogil Mole Ratio Key Free and Paid eBooks
 - Pogil Mole Ratio Key Public Domain eBooks
 - Pogil Mole Ratio Key eBook Subscription Services
 - Pogil Mole Ratio Key Budget-Friendly Options
6. Navigating Pogil Mole Ratio Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Pogil Mole Ratio Key Compatibility with Devices
 - Pogil Mole Ratio Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil Mole Ratio Key
 - Highlighting and Note-Taking Pogil Mole Ratio Key
 - Interactive Elements Pogil Mole Ratio Key
8. Staying Engaged with Pogil Mole Ratio Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Pogil Mole Ratio Key
- 9. Balancing eBooks and Physical Books Pogil Mole Ratio Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Mole Ratio Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pogil Mole Ratio Key
 - Setting Reading Goals Pogil Mole Ratio Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pogil Mole Ratio Key
 - Fact-Checking eBook Content of Pogil Mole Ratio Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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