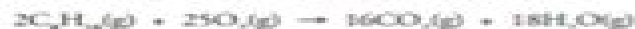


Extension Questions

Model 2 – Combustion Reactions



(C_8H_{18} = octane—gasoline component)



(C_4H_{10} = butane—lighter fuel)



(CH_4 = methane—natural gas)

19. What are the products in all of the combustion reactions in Model 2?

CO_2 and H_2O

20. What reactant is common in all of the combustion reactions in Model 2?

O_2

21. The "fuel" in most combustion reactions is a **hydrocarbon**. Using the examples in Model 2, write a description for the classification of substances known as hydrocarbons.

compound containing H + C

22. Predict the products of the following combustion reactions, and balance each reaction equation.



23. Compounds such as isopropyl alcohol ($\text{C}_3\text{H}_7\text{OH}$) and glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) can also undergo combustion in a similar way. Predict the products of the following combustion reactions, and balance each reaction equation.



24. For each of the reaction types in Model 1, explain why a combustion reaction CANNOT be classified as that type. Use a complete sentence in each answer.

S → combustion reactions don't just have 1 product

D → "

" have more than 1 reactant

SR → "

" need O_2 in both products not just 1

DR → "

" start w/ a compound and an element
not 2 compounds

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