

POGIL Activity – Gas Variables

(Teacher Notes)

- Assessment Questions and Responses:

Assessment Questions

1. Select the equation below that gives the correct relationship between the internal pressure of a gas and the volume of a flexible container.
 - a. $P = k/r$
 - b. $P = kV$
 - c. $V = kP$
 - d. $k = p/V$
2. Explain, on the molecular level, the change in internal pressure of a gas sample in a nonflexible container when the temperature is raised.
3. Explain, on the molecular level, the change in volume in a flexible container when more gas is added.

Assessment Target Responses

1. a.
2. *When the temperature of a gas is increased, the average kinetic energy of the molecules increases. This makes the gas molecules hit harder and more often, which increases the pressure.*

Pogil Gas Variables Packet Answers

C. E. R. Bruce



Pogil Gas Variables Packet Answers:

This book delves into Pogil Gas Variables Packet Answers. Pogil Gas Variables Packet Answers is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Pogil Gas Variables Packet Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Pogil Gas Variables Packet Answers
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This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Pogil Gas Variables Packet Answers.

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