

PHYSICAL SCIENCES P.2

MEMO-2024



basic education

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SENIOR CERTIFICATE

Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction in the above diagram that is a redox reaction.



4.8 Write the half-cell reaction for the anode in the above cell.

4.9 Write the overall cell reaction for the above cell.

5.1 Write down the name of the reaction in the above diagram.

5.2 Write down the name of the reaction in the above diagram.



5.3 Write down the name of the reaction in the above diagram.

5.4 Write down the name of the reaction in the above diagram.

5.5 Write down the name of the reaction in the above diagram.

5.6 Write down the name of the reaction in the above diagram.

Physical Science Paper Supplementary Memo 24

Karin Nielsen-Saines



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Parliament. House of Commons, 1916 Nuclear Science Abstracts , 1971-05 **Parliamentary Papers** Great Britain.
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Particle Physics Harvey B. Newman, Thomas Ypsilantis, 2012-12-06 The International Conference on the History of Original
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The most recent generation of experiments at e e and proton proton colliders and corresponding advances in theoretical
calculations have given us remarkably precise determinations of the basic parameters of the electroweak and strong
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our view of the many unanswered questions which remain for the next generation the origin and pattern of particle masses

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