

Separation of a Dye Mixture Using Chromatography

Lab Experiment Part 1

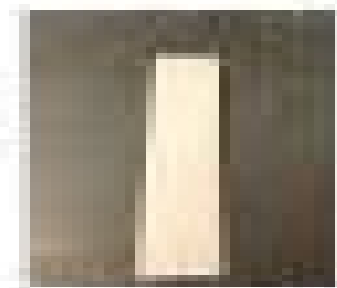
Objectives

1. Understand the principles of chromatography.
2. Perform a thin-layer chromatography (TLC) experiment.
3. Identify the components of a dye mixture.
4. Calculate the retention factor (R_f) for each component.
5. Compare the results with theoretical values.
6. Discuss the factors affecting the separation.
7. Prepare a chromatogram.
8. Interpret the chromatogram.
9. Write a lab report.
10. Present the results.

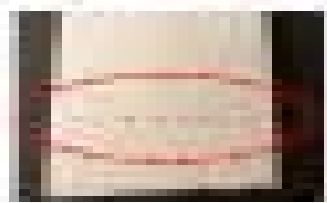
Introduction

Chromatography is a technique used to separate mixtures into their individual components. It is based on the principle of differential partitioning between a stationary phase and a mobile phase.

1. The stationary phase is a solid material, such as silica gel, coated on a glass plate or paper.
2. The mobile phase is a liquid or gas that carries the sample components through the stationary phase.



3. The separation occurs as the mobile phase moves through the stationary phase, carrying the sample components with it. The components are separated based on their affinity for the stationary phase.
4. The distance traveled by each component is measured, and the retention factor (R_f) is calculated.



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