

4. Complete the following table for the two experiments in Model 1.

Temp.	Experiment A	Experiment B
Independent Variable	# of particles	Temperature
Dependent Variable	Internal Pressure	Internal Pressure
Controlled Variables	External Pressure / Volume / Temperature	External Pressure / Volume / # of particles

5. Of the variables that were controlled in both Experiment A and Experiment B in Model 1, one requires a nonflexible container. Name this variable, and explain why a nonflexible container is necessary. In your answer, consider the external and internal pressure data given in Model 1.

Experiment Requires a nonflexible container as the independent variable is # of particles which would normally expand a container. The second experiment does not require a nonflexible container as the # of particles does not change. $\uparrow$ temp $\uparrow$ Volume

Read This!

Pressure is caused by molecules hitting the sides of a container or other objects. The pressure changes when the molecules change *how often* or *how hard* they hit. A nonflexible container is needed if the gas sample is going to have an internal pressure that is different from the external pressure. If a flexible container is used, the internal pressure and external pressure will always be the same because they are both

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