

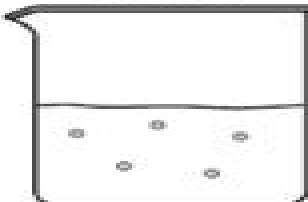
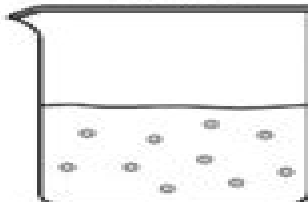
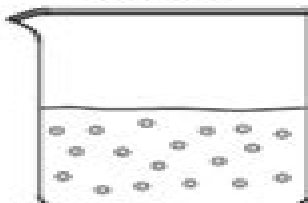
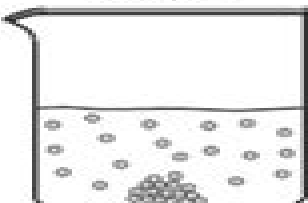
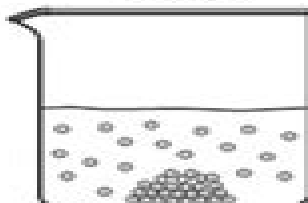
Saturated and Unsaturated Solutions

Is there a limit to the amount of solute that will dissolve in a solvent?

Why?

We use solutions every day. People who wear contact lenses use "lens solution" to rinse their contacts and keep them wet. Athletes who consume sports drinks after exercising benefit from the electrolytes in those solutions. This activity will explore whether or not there is a limit to how much of one substance can dissolve in another.

Model 1 – Saturated and Unsaturated Solutions

- All beakers contain 10.0 g of water. - All beakers are kept at 20 °C. - All solutions are stirred for 2 hours. - Solute is the same substance in all beakers.	Unsaturated Solutions	
	Beaker A  1.0 g of solute added	Beaker B  2.0 g of solute added
	Number of dissolved particles	5
	Number of solid particles	0
Saturated Solutions		
Beaker C  3.6 g of solute added	Beaker D  7.0 g of solute added	Beaker E  9.0 g of solute added
		18
	17	

Pogil Answer Key Solubility

E Staff



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Eventually, you will unconditionally discover a additional experience and exploit by spending more cash. yet when? complete you say yes that you require to acquire those all needs when having significantly cash? Why dont you try to get something basic in the beginning? Thats something that will guide you to comprehend even more re the globe, experience, some places, later than history, amusement, and a lot more?

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