



Name _____ Date _____

Investigation: The Properties of Water

Water, H_2O , is a polar molecule. The oxygen atom shares the pair of electrons with hydrogen in an unequal manner. Since oxygen is more electronegative than hydrogen, it has the tendency to pull the electrons towards itself more than hydrogen does. Consequently, a partial charge results at each end of the water molecule. Oxygen will have a negative partial charge and hydrogen will be slightly positive.



Water is a polar molecule, meaning it has one end with a slight positive charge and another end with a slightly negative charge. Molecules without positive and negative ends are called nonpolar. As a general rule, water is good at dissolving polar and ionic compounds, but does not dissolve nonpolar compounds.

These partial charges result in the ability of water to exhibit an intermolecular force, called a hydrogen bond, a weak yet important attraction between the hydrogen of one water molecule and the oxygen of a neighboring water molecule. H-bonds are important because they allow water molecules to:

1. Stick to other water molecules = cohesion
2. Stick to surfaces = adhesion
3. Absorb large amounts of heat before changing phase or boiling – high latent heat
4. Dissolve numerous polar and charged substances – good solvent

When water sticks to something, we call this adhesion. When you step out of the shower and see tiny droplets of water on your skin, this is an example of adhesion. In a plant, cohesion and adhesion help the plant by allowing water to travel upwards away from the roots to deliver water to all parts of the plant.

Water has a lower density in its solid form (ice) due to the way hydrogen bonds orient as they freeze. Water molecules actually push farther apart when they freeze, which causes frozen ice to float above the more dense liquid water. This phenomenon is important for living creatures to exist in water. The lower density of warm water is also responsible for ocean currents.

Station 1: Pennies and Water

Date	Tap Water	Soapy Water
Number of Drops		



1. What property or properties of water is demonstrated by this activity?

2. Explain why you had different results for the soapy water.

Pogil Activities For Ap Biology Properties Of Water

Jiyuan Zhang



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