

Physical Science
Chapter 2

Properties of Matter

Classifying Matter

Substances

- **Element:** a substance that can't be separated down into simpler substances.
- **Atom:** the smallest particle of an element.
- An element has a fixed composition because it contains only one type of atom.

2 Types of Mixtures

1. **Heterogeneous Mixture:** have materials that are not evenly distributed and noticeably different from one another.
 2. **Homogeneous Mixture:** have materials that are evenly distributed so it is difficult to distinguish one substance from another.
 - a. Also called a solution
 - b. Remain constantly and uniformly mixed
- Which type of mixture is milk?
 - Based on the size of its largest particles, a mixture can be classified as a solution, a suspension, or a colloid.

Solutions

Solution: when substances dissolve and form a homogenous mixture.

- Examples: windshield wiper fluid, sea water, lemonade

Colloids

Colloid: contains some particles that are larger than those in a solution, but smaller than those in a suspension.

- Examples: homogenized milk, fog
- Think about it...
 - What happens to the size of the drops of fat in milk when milk is homogenized?

The Scattering of Light

- Colloids and Suspensions scatter light just like fog.
- Light goes right through a solution.

Chemical Properties

- Any ability to produce a change in the composition of matter
- Chemical properties can be observed only when the substance in a sample of matter is changing into different substances.

Physical Science Chapter 2 Review Answers Properties Of Matter

J Elliott



Physical Science Chapter 2 Review Answers Properties Of Matter:

Reviewing **Physical Science Chapter 2 Review Answers Properties Of Matter**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Physical Science Chapter 2 Review Answers Properties Of Matter**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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