

Off-priming and Concomitancy Issues 1 2009

1000

89. During a chemical reaction, the activation energy is the energy difference between the reactants and the products. What is the activation energy for the reaction below?
- a. 100 kJ/mol
b. 200 kJ/mol
c. 300 kJ/mol
d. 400 kJ/mol
e. 500 kJ/mol
90. Which of the following is a good example of a chemical reaction?
- a. Boiling water
b. Dissolving salt in water
c. Mixing oil and water
d. Combustion of gasoline
e. None of the above
91. Which of the following is a good example of a chemical reaction?
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