



basic education

Department:
Basic Education
REPUBLIC OF SOUTH AFRICA

SENIOR CERTIFICATE/
NATIONAL SENIOR CERTIFICATE

GRADE 12

ENGINEERING GRAPHICS AND DESIGN P1

NOVEMBER 2020

MARKS: 100

TIME: 3 hours

This question paper consists of 6 pages.

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INSTRUCTIONS AND INFORMATION

1. This question paper consists of FOUR questions.
2. Answer ALL the questions.
3. All drawings are in first-angle orthographic projection, unless otherwise stated.
4. All drawings must be prepared using pencil and instruments, unless otherwise stated.
5. All answers must be drawn accurately and neatly.
6. All the questions must be answered on the QUESTION PAPER, as instructed.
7. All the pages, irrespective of whether the question was attempted or not, must be re-stapled in numerical sequence in the TOP LEFT-HAND CORNER ONLY.
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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^{-} + \text{Au} + \text{CN}^{-} + 2\text{e}^{-} + 2\text{H}^{+} + \text{H}_2\text{O}_2 \rightarrow 2\text{HCN} + 2\text{H}_2\text{O}$ make electrons equal convert to basic solution $4\text{CN}^{-} + 2\text{Au} + 2\text{CN}^{-} + 2\text{e}^{-}$ multiplied by a factor of 2 $2\text{H}_2\text{O}_2 + \text{O}_2 + \text{H}_2\text{O}_2 + 2\text{OH}^{-}$ 3 the final answer

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web balancing redox reactions in acidic solution problems 1 10 problem 1 $\text{Cr}_2\text{O}_7^{2-}$ Fe^{2+} Cr^{3+} Fe^{3+} solution 1 balanced half reactions 6e^- 14H^+ $\text{Cr}_2\text{O}_7^{2-}$ 2Cr^{3+} $7\text{H}_2\text{O}$ Fe^{2+} Fe^{3+} e 2 equalize the electrons 6e^- 14H^+ $\text{Cr}_2\text{O}_7^{2-}$ 2Cr^{3+} $7\text{H}_2\text{O}$ 6Fe^{2+} 6Fe^{3+} 6e^- multiplied by a factor of 6

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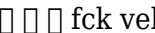

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