

Question	Answer	Mark	Comments
1	<p> $E = hf$ $E = 6.63 \times 10^{-34} \times 5.0 \times 10^{14}$ $E = 3.315 \times 10^{-19} \text{ J}$ $E = 2.07 \text{ eV}$ </p>	<p>100</p> <p>100</p>	
2(a)	<p> Increased frequency through the LED </p>	100	100 mark band for appropriate full explanation
2(b)	<p> $E = hf$ </p>	100	
2(c)	<p> $E = hf$ $E = 6.63 \times 10^{-34} \times 5.0 \times 10^{14}$ $E = 3.315 \times 10^{-19} \text{ J}$ $E = 2.07 \text{ eV}$ $E = 2.07 \text{ eV}$ </p>	<p>100</p> <p>100</p> <p>100</p> <p>100</p> <p>100</p>	
2(d)	<p> Increase in potential with a frequency increase </p>	100	
2(e)	<p> Increase in $E = hf$ with frequency increase Frequency increase </p>	<p>100</p> <p>100</p>	
2(f)	<p> $E = hf$ $E = 6.63 \times 10^{-34} \times 5.0 \times 10^{14}$ $E = 3.315 \times 10^{-19} \text{ J}$ </p>	<p>100</p> <p>100</p>	100 mark band
2(g)	<p> $E = hf$ </p>	100	100 mark band
2(h)	<p> Increased kinetic energy Increased frequency The photon energy gets increased through $E = hf$ $E = hf$ </p>	<p>100</p> <p>100</p> <p>100</p> <p>100</p>	

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