

Name: _____

Persuasive Writing CCS Aligned Rubric

CCS / Concept	4 (Above Grade Level)	3 (At Grade Level)	2 (Approaching Grade Level)	1 (Below Grade Level)
W.1.1 Name a topic	Names and introduces a topic	Clearly states a topic	Topic is not clearly stated	Topic is unclear
W.1.1 Supply reasons	Supply reasons that support an opinion	Supply a reason for an opinion	State an opinion but may not include a logical reason	Does not state an opinion or include a reason
W.1.1 Use an ending	Provides concluding statement or section	Provides sense of closure	Attempts to provide closure	Does not provide closure
W.1.2 Focus on a topic and strengthen writing	Focuses on a topic and strengthens writing as needed by revising and editing	Focuses on a topic and adds details to strengthen writing as needed	Attempts to focus on a topic and strengthen writing	Does not focus on a topic or attempt to strengthen writing
L.1.1 Uses upper- and lower-case letters	Capitalizes first word in a sentence, "I," and proper nouns correctly	Capitalizes first word in a sentence and "I"	Capitalizes first word in a sentence most of the time	Capitalizes inconsistently
L.1.1 Produce complete sentences	Produces, expands, and rearranges simple and compound sentences	Produces correct simple and compound sentences	Produces mostly correct simple and compound sentences	Produces mostly incorrect simple and compound sentences
L.1.1 Use frequently occurring adjectives	Choose between adjectives (describing words) to fit purpose	Use frequently occurring adjectives (describing words)	Use the most frequently occurring adjectives	Incorrectly or inaccurately uses adjectives
L.1.2 Spells sight words correctly	Applies conventional spelling for words with common spelling patterns and irregular sight words	Applies conventional spelling for words with common spelling patterns and frequently occurring irregular words	Applies conventional spelling for most common and short-vowel words	Applies little to no sound/ spelling correspondence of consonants and short vowels
L.1.2 Sentences end with punctuation	Uses end punctuation correctly all the time and exclamation, !, to create excitement.	Uses end punctuation correctly all the time	Uses end punctuation correctly some of the time	Uses end punctuation incorrectly or not at all
L.1.2 Includes all or most sounds in words	Spells with learned spelling patterns with untaught words	Spells untaught words phonetically	Spells simple words phonetically	Spells few to no simple words phonetically

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G Orfield



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$ $\Delta H = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g})$ $\Delta H = 283 \text{ kJ}$

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web chemistry 120 hess's law worksheet calculate ΔH for the reaction $\text{C}_2\text{H}_4(\text{g}) + \text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_6(\text{g})$ from the following data
 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l})$ $\Delta H = 1411 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l})$ $\Delta H = 1560 \text{ kJ}$
 $\text{H}_2(\text{g}) + \frac{1}{2} \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta H = 285 \text{ kJ}$

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web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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web thus for reaction 9 2 in which 1 mole of H_2O and 1 mole of OH^- combine to form 2 moles of H_2O $\Delta H = -55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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