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**Botha, Smuts and the Great War** Antonio Garcia, Ian van der Waag, 2023-10-30 In Botha Smuts and the Great War 1914 1918 authors Antonio Garcia and Ian Van Der Waag conducted painstaking research in South Africa and the United Kingdom to produce this first of a kind volume on the wartime roles of South African prime minister General Louis Botha and his deputy General Jan Smuts These very different men appealed to different audiences Botha's nuance and emotional intelligence complemented Smuts's intellectualism Thrown into a world conflagration in August 1914 and facing internal rebellion and the threat posed by German troops on the borders they led South Africa's Union Defence Force Both Botha and Smuts commanded in the field Steadily the South African army they commanded benefiting from wartime training sometimes in the field gained resilience experience and battle hardiness adapting to the conditions of the campaigns and the demands of the tasks South Africa's campaigns were complex and divergent starting with the invasion of neighbouring German South West Africa to neutralise enemy radio stations and so aid the security of the South Atlantic Suddenly suspended following the outbreak of the Afrikaner Rebellion the campaign recommenced in January 1915 Following its conclusion an infantry brigade raised for Western Front service was diverted to Egypt before facing near annihilation at Delville Wood Simultaneously a large South African force fighting alongside British African and Indian forces overcame German resistance in East Africa whilst a brigade of field artillery and later the Cape Corps served in Egypt and Palestine Moreover approximately 6 500 South Africans served in the British Army Royal Flying Corps Royal Air Force and in the Royal Navy Although lionised during the war by a British public hungry for heroes there is a different side to Botha and Smuts Shunned by Afrikaner nationalists at the time they have remained divisive figures Responsible for the enactment of the Land Act of 1913 which shaped South Africa's socio economic and political landscape Botha's statues in Cape Town Durban and Pretoria were vandalised on a number of occasions between 2015 and 2022 and there were recent calls for Smuts's statues to be removed Behind Botha's charming attractive fa

ade and Smuts's stoic machine were two very human imperfect and quite probably inconsiderate men Together they provide a wonderful lens through which to examine the potent forces of the early twentieth century world and the country they hoped to forge Myopic compatriots had constrained their plans but it was the outbreak of war in 1914 that offered the most significant opportunities and brought the most adverse challenges They fought insurmountable odds and achieved great victories at home and abroad but also made startling errors and ultimately in classical fashion risked being crushed by the weight of the world they tried to create

**The SA Navy's New Acquisitions and Some Thoughts on Naval Presence**

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web calculate the mass percent mole fraction molality and molarity of hexane  $\text{C}_6\text{H}_{14}$  d 0 660 g  $\text{cm}^3$  when it is mixed with heptane  $\text{C}_7\text{H}_{16}$  d 0 684 g  $\text{cm}^3$  in a 35 0 55 0 ml ratio assume that the volumes are additive

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web aug 30 2012 here we ll do practice problems with molarity calculating the moles and liters to find the molar concentration we ll also have to use conversion factors to convert between grams and moles and

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web for the first five problems you need to use the equation that says that the molarity of a solution is equal to the number of moles of solute divided by the number of liters of solution in this problem simply solve using the molarity equation to find that the concentration of the solution is 10 m

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web problem 6 1 6 1 6 calculate the molarity of each of the following solutions a 0 195 g of cholesterol  $\text{C}_{27}\text{H}_{46}\text{O}$  in 0 100 l of serum the average concentration of cholesterol in human serum b 4 25 g of  $\text{NH}_3$  in 0 500 l of solution the concentration of  $\text{NH}_3$  in household ammonia

**15 03 solution concentration molality mass percent ppm and** - Jul 20 2023

web a similar unit of concentration is molality m which is defined as the number of moles of solute per kilogram of solvent not per liter of solution molality  $\frac{\text{moles solute}}{\text{kilograms solvent}}$  mathematical manipulation of molality is the

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web 1 molarity mv mass molar mass x 0 4500 l 0 825 g 141 9579 g mol x 0 0129 m 2 molality 0 825 g 141 9579 g mol 0 00581158 mol 0 00581158 mol 0 4500 kg 0 0129 m 3 mole fraction na 2 hpo 4 0 825 g 141 9579 g mol 0 00581158 mol h 2 o 450 0 g 18 015 g mol 24 97918401 mol

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web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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web what mass of  $\text{CoCl}_2$  in grams is needed for the solution express the answer using 3 significant figures

**8 3 concentrations of solutions problems chemistry libretexts** - Dec 13 2022

web the hardness of water hardness count is usually expressed in parts per million by mass of  $\text{CaCO}_3$  which is equivalent to milligrams of  $\text{CaCO}_3$  per liter of water what is the molar concentration of  $\text{Ca}^{2+}$  ions in a water sample with a hardness count of 175 mg  $\text{CaCO}_3$  l

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web answer the mass percentage of an element of a solution is defined as the mass of the solute in grammes present in 100 g of the solution it s written like this 
$$\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$$
 end array

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web aug 13 2017 molarity practice problems this chemistry video tutorial explains how to solve common molarity problems it discusses how to calculate the concentration of a solution given the mass in grams