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NO MASTER

BOUNTY - BOOK THREE



No Master Bounty Three

Mary Victoria Cowden CLARKE



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web equilibrium moles $hscn\ 2.00 \times 10^{-5}\ mol\ 3.00 \times 10^{-6}\ mol\ 1.70 \times 10^{-5}\ mol\ hscn$ knowing the number of moles of fe^{3+} and $hscn$ present in the equilibrium mixture and the volume of the mixture we can easily find the concentrations of those two species $mol\ fe^{3+}\ 1.70 \times 10^{-5}\ mol\ fe^{3+}\ 8.50 \times 10^{-4}\ m$

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