

- 10. The reaction is exothermic because the products have a lower energy than the reactants.
- 11. The activation energy is the energy difference between the reactants and the products.
- 12. The energy of the reactants is higher than the energy of the products.
- 13. The energy of the products is lower than the energy of the reactants.



Figure 1



Figure 2

Figure 1 shows the energy profile for an exothermic reaction. Figure 2 shows the energy profile for an endothermic reaction.

Answer the questions in the margin.

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