# PRACTICING

Pema Chodron

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**ML Morrison** 

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web all reactions tend towards a state of chemical equilibrium the point at which both the forward process and the reverse process are taking place at the same rate since the forward and reverse rates are equal the concentrations of the reactants and products are constant at equilibrium

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web equilibrium the rate of the forward reaction and the rate of the reverse reaction are equal at equilibrium both reactants and products are present some reaction mixtures contain mostly reactants and form only a few products at equilibrium some reaction mixtures contain mostly products and remain mostly reactants at equilibrium

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web chemical equilibrium part 1 forward and reverse reactions understanding the concept of chemical equilibrium is critical to following several of the discussions that we have in bis2a and indeed throughout biology and the sciences