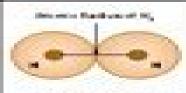
# Periodic Trends Practice



#### Alomic Radii

The distance from the nucleuses of two bonded atoms.



#### Electronegativity

How much an atom. "wants" electrons on the atom's fundancy to attract electrons.



# **Energy**

The amount of energy required to remove electrons from an atom.



#### Electron Affinity

The amount of energy released when a neatral gaseous atom gains an electron.



#### **Driving Question**

How can the periodic table be used to predict atomic characteristics such as atomic radius, electronogetivity, isnisation energy, and electron affinity.

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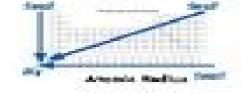
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Shielding?

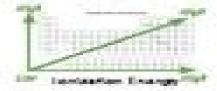
What is shielding? What effect does it have on the periodic trends defined above?

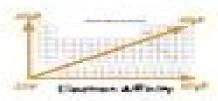
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Add arress to each of the following periods tables to indicate the direction of the trend as it increases or decreases from left to right and top to bettern. If the Dirichlet solds will be periods tobles so you can easily remember their









# **Practice Identifying Periodic Trends**

**Daniel F McAuley** 

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