

Group 15 Elements

N, P, As, Sb, Bi, Mc

Hydrides of Group 15 Elements

Halides of Group 15 Elements

The valence shell electronic configuration is ns^2np^3

N and P - non-metals

As and Sb - metalloids

Bi and Mc (moscovium) - metals

Common oxidation states are -3, +3 and +5

NH_3 has higher melting and boiling points than PH_3 due to hydrogen bonding

NH_3 is a mild reducing agent and BiH_3 is the strongest reducing agent

Basicity order - $NH_3 > PH_3 > AsH_3 > SbH_3 > BiH_3$

Pentahalides are more covalent than trihalides

Nitrogen does not form pentahalide due to absence of the d-orbitals in its valence shell

P Block Elements Class 12 Full Chapter

Disha Experts



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CBSE Class 12 Chemistry English Story Chapter 4 Priti Singhal,2024-11-10 This book is structured to align with the latest syllabus and curriculum guidelines ensuring the content is relevant and rigorous Each chapter begins with a clear set of learning objectives providing a roadmap for students to understand what they will achieve by the end of the chapter We have included numerous diagrams illustrations and real life examples to make complex concepts more accessible and engaging *10 in One Study Package for CBSE Chemistry Class 12 with 5 Model Papers* Disha Experts,2017-08-29 10 in ONE CBSE Study Package Chemistry class 12 with 5 Sample Papers is another innovative initiative from Disha Publication This book provides the excellent approach to Master the subject The book has 10 key ingredients that will help you achieve success 1 Chapter Utility Score 2 All India Board 2017 Solved Paper 3 Exhaustive theory based on the syllabus of NCERT books along with the concept maps for the bird s eye view of the chapter 4 NCERT Solutions NCERT Exercise Questions 5 VSA SA LA Questions Sufficient Practice Questions divided into VSA SA LA type Numericals are also included wherever required 6 Past Years Questions Past 10 year Questions of Board Exams are also included 7 HOTS Exemplar Value based Questions High Order Thinking Skill Based Moral Value Based and Selective NCERT Exemplar Questions included 8 Chapter Test A 15 marks test of 30 min to assess your preparation in each chapter 9 Important Formulae Terms and Definitions 10 Full syllabus Sample Papers 5 papers with detailed solutions designed exactly on the latest pattern of CBSE Board

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Content 1 Some Basic Concepts of Chemistry 2 Structure of Atom 3 Classification of Elements and Periodicity in Properties 4 Chemical Bonding and Molecular Structure 5 States of Matter 6 Thermodynamics 7 Equilibrium 8 Redox Reactions 9 Hydrogen 10 s Block Elements 11 p Block Elements 12 Organic Chemistry Some Basic Principles and Techniques 13 Hydrocarbons 14 Environmental Chemistry I

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