

Pre- AP Chemistry ~ Unit 6 – Notes Packet - chemical reactions

- **Chemical Reaction Definition:** the process by which one or more substances are changed into one or more different substances.
- How do we know a chemical change has occurred?
 - Evidence that a chemical change *may have occurred* can include:
 - Production of bubbles (gas)
 - Formation of a precipitate (solid)
 - Production of heat and light
 - Color change
 - Noise
 - Ultimately, only the formation of a NEW substance can be absolute proof that a chemical change has occurred.
- How do we write chemical reactions?
 - Include multiple molecules
 - Reactants $\longrightarrow$ products
 - (original substances) (resulting substances)
- How Do Reactants Turn Into Products?
 - Electrons from atoms want to get into the lowest energy state possible
 - They will rearrange themselves with other atoms to do this
 - Reactants come into contact and collide
 - Bonds between reactant atoms break
 - This gives freedom to atoms to rearrange
 - Atoms form bonds with new atoms to create products
- Reactions and Energy
 - One of the keys to making reactions “go” is energy:
 - Energy is the ability to do work (to move or change matter)
 - Most reactions will either release or absorb energy

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