

* Plane of Symmetry :-



→ Planar molecule

① CH₄



Gh ⊗
Ev ⊗

②



Gh ⊗

Ev ⊗

③



Gh ⊗

Ev ⊗

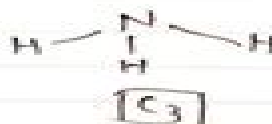
* Axis of Symmetry :-

C₂, C₃, C₄, C₅, C₆ (C_n) :-

① CH₄



② NH₃



$$n = \frac{360}{9}$$

$$n = \frac{360}{120} = 3$$

Practice For Stereo Chemistry

L Manion



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