

Name: _____

Class: _____

OXIDATION NUMBERS WORKSHEET

The *oxidation number* of an atom is the apparent charge assigned to it in a particular molecule, ion or compound. Certain rules are followed in assigning oxidation numbers.

Use rules at the right to assign oxidation numbers to each element in each of the given formulas.

Example

	OXIDATION NO.	RULES USED
H_2O	<u>H=+1, O= -2</u>	<u>6 & 7</u>
N_2	<u>N= 0</u>	<u>1</u>

Rules The oxidation number of:

1. A free element in the uncombined state is 0.
2. A monatomic ion equals the charge on the ion.
3. Oxygen in a compound is generally -2
4. Hydrogen in a compound is +1
5. In a neutral compound the sum of the oxidation numbers for all the atoms is 0.
6. In a polyatomic ion the sum of the oxidation numbers for all the atoms is equals the charge on the ion.
7. Group 1 metals is +1.
8. Group 2 metals is +2.

		RULES USED
1.	Cl_2	1. _____
2.	Cl^-	2. _____
3.	Na	3. _____
4.	Na^+	4. _____
5.	KCl	5. _____
6.	H_2S	6. _____
7.	CaO	7. _____
8.	H_2SO_4	8. _____
9.	NO_3^-	9. _____
10.	$\text{Cr}_2\text{O}_7^{2-}$	10. _____
11.	NH_4Cl	11. _____
12.	NH_3	12. _____
13.	NO_2	13. _____
14.	CaH_2 (calcium hydride)	14. _____
15.	Na_2O_2 (sodium peroxide)	15. _____

Oxidation Number Practice Answer Key

Julia Burdge



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for JEE Advanced: Part 1, 3E (Free Sample) K. S. Verma, 2022-05-19 Physical Chemistry for JEE Advanced Part 1 a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of physical chemistry from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers

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